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S.Sudo, F. Sano, H. Zushi, T. Mizuuchi, S. Besshou, H. Okada,
K. Nagasaki, K. Sakamoto, Y. Kurimoto, H. Funaba,
T. Hamada, T. Kinoshita, S. Kado, Y. Kanda,
T. Okamoto, M. Wakatani and T. Obiki

(Received - June 5, 1996)

NIFS-421

July 1996

# RESEARCH REPORT NIFS Series

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## Behavior of Pellet Injected Li Ions into Heliotron E Plasmas

K.Kondo<sup>1)</sup>, K.Ida<sup>3)</sup>, C.Christou<sup>1)</sup>, V.Yu.Sergeev<sup>4)</sup>, K.V.Khlopenkov<sup>4)</sup>, S.Sudo<sup>3)</sup>, F.Sano<sup>2)</sup>, H.Zushi<sup>1)</sup>, T.Mizuuchi<sup>2)</sup>, S.Besshou<sup>1)</sup>, H.Okada<sup>2)</sup>, K.Nagasaki<sup>2)</sup>, K.Sakamoto<sup>2)</sup>, Y.Kurimoto<sup>5)</sup>, H.Funaba<sup>5)</sup>, T.Hamada<sup>5)</sup>, T.Kinoshita<sup>5)</sup>, S.Kado<sup>6)</sup>, Y.Kanda<sup>5)</sup>, T.Okamoto<sup>5)</sup>, M.Wakatani<sup>1)</sup> and T.Obiki<sup>2)</sup>

<sup>1)</sup>Faculty of Energy Science, Kyoto University, Uji, Japan
 <sup>2)</sup>Institute of Advanced Energy, Kyoto University, Uji, Japan
 <sup>3)</sup>National Institute for Fusion Science, Nagoya, Japan
 <sup>4)</sup>St.Petersburg Technical University, Russia

<sup>5)</sup>Faculty of Engineering, Kyoto University, Kyoto, Japan

<sup>6)</sup>Interdisciplinary Graduate School of Engineering Science, Kyushu University, Kasuga, Japan

#### abstract

Li pellet injection has provided a complex plasma with a large fraction of Li ions, which is characterized by intense emissions from Li I and III. The spatial profiles of the fully ionized Li<sup>3+</sup> ions are measured by charge exchange recombination spectroscopy with a resolution of 13 mm, and the local decay time of the injected Li ion has been estimated. The spectral profile of the charge exchange recombination line of Li<sup>2+</sup> from n=5 to n=4 shows a complicated structure, which depends on Li<sup>3+</sup> density. The effects on other intrinsic impurities and recycled Li are also discussed.

Key Words: lithium pellet injection, Heliotron E, charge exchange recombination spectroscopy, impurity particle behavior, spectral profile of Li<sup>2+</sup> ions,

#### 1. Introduction

Li pellet injection has been recently in producing high performance plasmas [1,2,3]. The ablation mechanism of Li pellet is also of interest because of high sublimation energy compared to hydrogen and deuterium pellets [4,5]. The development of Li pellet injectors and study of the interaction between injected Li and plasma are thus urgent issues for fusion oriented experiments. In Heliotron E [6], a Li-pellet injector has been developed in order to obtain good performance plasmas, to study the ablation mechanisms and to establish a new diagnostic method for the measurement of the magnetic field direction. Injected Li is also used as a tracer for investigating particle transport and for analyzing spectral profiles of Li ions. It is important to study the behavior of the injected Li ions for understanding above subjects. In this paper, the behavior of the injected Li ions in neutral beam heated plasmas is presented. In paticular fully ionized Li<sup>3+</sup> ions are measured by a charge exchange recombination spectroscopy, and the local decay time of Li<sup>3+</sup> is estimated. Effects of Li on intrinsic impurities are also described.

## 2. Li pellet injector and diagnostics

The Li pellet injector consists of a magazine for pellets, gun barrel, guide tubes and differential pumping system with a buffer tank of 0.1 m<sup>3</sup> [7]. 30 pre-shaped Li pellets 1mm in diameter and length are filled into the holes in the rotatable disk of the magazine. A single pellet has  $3.6 \times 10^{19}$  Li atoms. The gun barrel is 1 m long and 1 mm in diameter. The four guide tubes are 0.14 m, 1.5 m, 0.4 m, and 2.7 m long, respectively. The diameters of these guide tubes are 2.27 mm, 4 mm, 7.53 mm and 10.7 mm, respectively. High pressure He is used as a propellant gas. The injection speed is 400~500 m/s, which is measured between the gun barrel and the first guide tube. The pellet trajectory distributes within a cone of 1 degrees full angle. Pellets are injected horizontally and cross the plasma column nearly perpendicularly. The minor radius of the plasma column is 0.3 m, and the ablation time is 1.3 ms. A high speed camera is used to record images of the ablation cloud at every 50 µs. Two optical filters for Li I (6706 Å) and Li II (5485 Å) are provided to estimate the ablation rates. A multi-channel vacuum ultraviolet spectrometer, which is available from 45Å to 400Å, is used to measure Li III (135) Å) emission. The neutral Li I (6706 Å) emitted near the wall is measured by a visible spectrometer, which is separated by 132 degrees in the toroidal direction from the port of the pellet injector. A 1.26 m visible spectrometer with a 2-dimensional detector

(Hamamatsu,C3554X) [8] is employed to analyze the spectral profiles of the Li<sup>2+</sup> line populated by charge exchange recombination from Li<sup>3+</sup> ion. The time resolution is 20 ms. The sight-lines of 40 fiber optics cross the plasma column from 1.98 m to 2.42 m on a major radius and the spatial resolution is 13 mm. The profiles of the electron density and temperature are measured by a Thomson scattering system [9], and ion temperature profile is estimated by the above charge exchange recombination spectroscopy.

#### 3. Behavior of Li ions

A Li pellet is injected into a plasma, which is initiated by a 106GHz gyrotron and further heated by neutral beams of 3.5MW power. The acceleration voltage of the neutral beams is 23 kV. The magnetic field strength is 1.9 T. The vacuum chamber wall has been "boronized" by using an ECH discharge with  ${\rm B}_{10}{\rm H}_{14}$  and He [10]. Figure 1 shows the time evolutions of the average electron density with and without Li pellet. The incremental electron density by pellet injection is also shown. The plasma is produced at 275 ms and the neutral beam heating starts at 300 ms and terminates at 444 ms. After the Li pellet is injected at 327 ms, the average electron density increases by  $4\times10^{13}$  cm<sup>-3</sup>. Figure 2 (a) and (b) show the profiles of the electron temperature and density immediately before and after the Li pellet injection. The central electron temperature drops from 0.75 keV to 0.45 keV, and the central electron density increases from  $4.5 \times 10^{13}$  cm<sup>-3</sup> to  $1.4 \times 10^{14}$  cm<sup>-3</sup>. The electron energy has almost doubled after the Li pellet injection. The ablation rate of 3.5×10<sup>22</sup> atoms/s is estimated at the plasma center. In this case, the increase of the electron density is smaller than that provided by fully ablated Li pellet. The injected Li pellet does not completely ablate and about 60% of the pellet is deposited in the plasma. The density ratio of the Li<sup>3+</sup> ions to protons is estimated to be 0.6, but this ratio decreases rapidly as shown later.

Figure 3 shows spectra in the wavelength region from 170 Å to 400 Å before (320 ms) and after (330 ms) the Li pellet injection. The integration time is 10 ms. The base line is shifted in the upper spectrum. The prominent line is Li III (135 Å), which appears at 270 Å in the second order. The emission of Li II (199.3 Å) is not observed. Other prominent lines originate from intrinsic impurities such as carbon, oxygen, titanium, chromium and iron. Figure 4 shows the time evolution of the Li III, O V(192.9 Å) and Fe XV (284.1 Å) emissions are also shown as typical examples of light and heavy impurities. Immediately after Li pellet injection emission

of Li III peaks and decreases with a decay time of about 60 ms. Similarly a sharp peak appears in Fe XV emission, which increases again from 360 ms to the end of the neutral beam pulse due to continuous influx of iron. Pellet injection causes emission from the high ionization states of metallic impurities to increase significantly, but appears to have a much smaller effect on oxygen and carbon emission. This suggests that the pellet penetrates into the plasma core and iron emission may be enhanced by either an increase in central electron density or by recombination of higher ionization states caused by a drop of the electron temperature.

The fully ionized  ${\rm Li}^{3+}$  ion has been studied using charge exchange recombination spectroscopy. The transition from n=5 to n=4 is selected, the only one available to our diagnostic system. The wavelength from  ${}^2{\rm G}$  to  ${}^2{\rm F}$  is 4499 Å [11]. Figure 5 shows the intensity profiles of 4499 Å from 370 ms to 450 ms. The horizontal axis is the major radius and the magnetic axis is located at 2.2 m. The Li pellet is injected at 378 ms. The decay time of  ${\rm Li}^{3+}$  is estimated at various locations by exponential fitting from 390 ms to 430 ms. At R=2.2 m, on the magnetic axis, the decay time is 90 ms, and 160 ms is obtained at R=2.1 m and 60 ms at R=2.3 m. There appears an asymmetry between inboard and outboard of the plasma column. This asymmetry is attributed to the slow increase of the penetrated neutral beam density in the inboard due to the decrease in the electron density as described later. The Li III (135 Å), observed by the VUV spectrometer, shows a decrease with a decay time of 60 ms. In this shot the average electron density increased from  $2.5 \times 10^{13}$  cm $^{-3}$  to  $9 \times 10^{13}$  cm $^{-3}$  by the pellet, and then decreased to  $7 \times 10^{13}$  cm $^{-3}$  with a decay time of 160 ms.

Figure 6 shows the spectral profiles of around 4499 Å at sucessive times after the pellet. There appear to be several line components, which are in the wavelength region corresponding to transitions from the n=5 sublevels;  $^2$ P,  $^2$ D,  $^2$ F and  $^2$ G. Arrows in the figure indicate the calculated average wavelengths from designated upper levels. The shorter wavelength components are dominant in the early phase after the Li injection. As Li density decreases, the dominant component shifts to the longer wavelength. This corresponds to the transition from low orbital angular momentum to high orbital angular momentum transitions. The depopulation among the levels due to collision in high density regime is considered as a candidate for the explanation, but there is no similar phenomenon in carbon. In the Li<sup>3+</sup> ion case, the density is so large it is possible that this depopulation is due to collisions among Li ions.

#### 4. Behavior of the neutral Li line

Neutral Li emission at 6706 Å is used to study the behavior of recycled Li after the Li pellet injection. A spectrometer with a focal length of 1.26 m and a 1024 channel diode array detector is used. The time resolution is 10 ms. Figure 7 shows the shot-to-shot variations of Li I intensity. Li shots are denoted by closed circles and open squares correspond to shots without a pellet. The neutral beam power is 3.2~3.4 MW. For the shots with pellet the intensities are estimated at the time before the Li pellet is injected. The Li I emission is again visible from shots following pellet injection shots indicating Li recycling. Injection of Li pellets in successive shots can cause a cumulative increase in Li I emission and a gradual fall in Li I emission can be seen one to two shots after Li pellet injection has ceased. When corrected for differences in electron density, the shot-to-shot decay of Li I density appears to be exponential. Oxygen is an interesting impurity in pellet injected shots. The intensity typically decreases in successive shots, but no reduction in the relative impurity level normalized by the electron density is observed.

#### 5. Conclusion

In Heliotron E, a Li pellet injector with good reproducibility and reliability has been developed and successfully injected into neutral beam heated plasmas. The spatial profiles of the Li III emission from Li<sup>3+</sup> are measured with a resolution of 13 mm, and local decay time is estimated. This diagnostic method can be used to estimate the local values. Moreover the emission intensity is determined only by the product of neutral beam and Li<sup>3+</sup> densities so complicated analysis on excitation and ionization is not necessary. The fine structure from sublevels of the charge exchange recombination line from n=5 to n=4 is analyzed. The collision among Li ions is considered to be the most probable candidate for the depopulation.

### Acknowledgment

The authors are grateful to the Heliotron group for their excellent and helpful support. The authors would like to thank Dr. J. Terry for critical reading of the manuscript.

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## Figure captions

Fig.1

The time evolutions of the average electron densities. The solid line is for the shot with Li pellet injection at 336 ms and the dotted line corresponds to the shot without pellet injection. The incremental density by pellet injection is also plotted.

Fig.2

The electron temperature (a) and density (b) profiles immediately before(closed) and after(open) the pellet injection.

Fig.3

The spectra in the vacuum ultraviolet region before and after the pellet injection. Li III(135 Å) appears at 270 Å in second order.

Fig.4

The time evolutions of the intensities of Li III, O V and Fe XV.

Fig.5

Spatial profiles of Li<sup>2+</sup> emissions measured by charge exchange recombination spectroscopy.

Li pellet is injected at 376 ms.

Fig.6

Spectral profiles of charge exchange recombination line from n=5 to n=4 transition. Arrows indicate the calculated average wavelengths from designated upper levels.

Fig.7

Li I intensity in successive shots with and without Li pellets. Closed circles are with Li pellet injection and open squares are without.

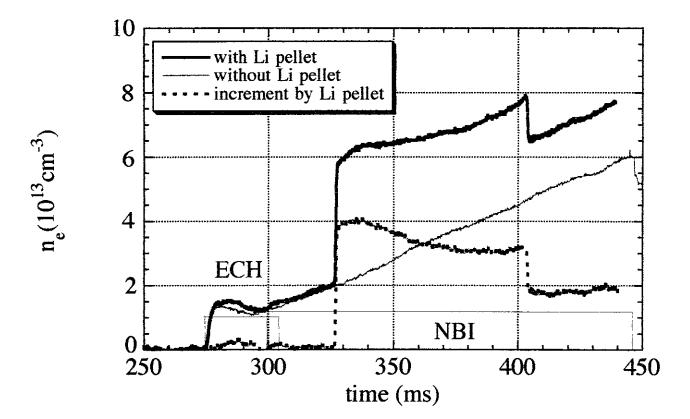


Fig.1

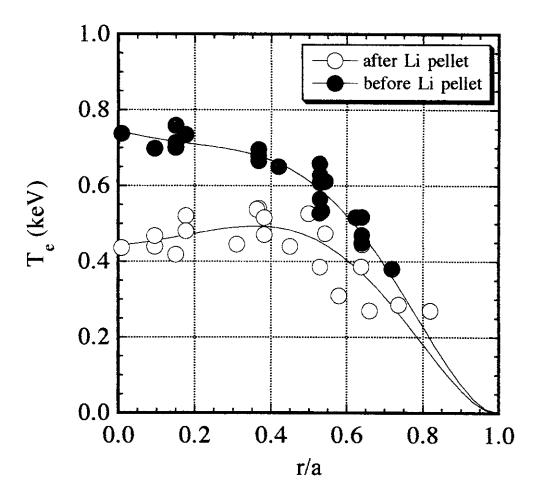


Fig.2(a)

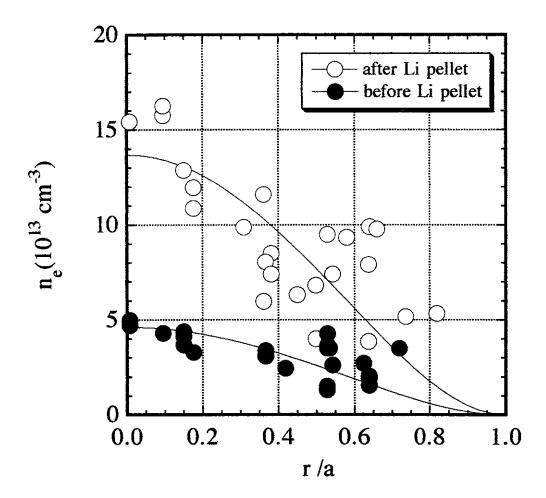


Fig.2(b)

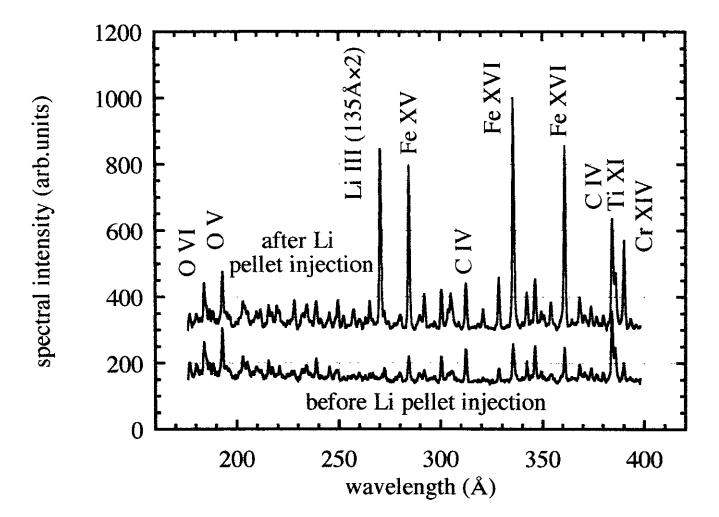


Fig.3

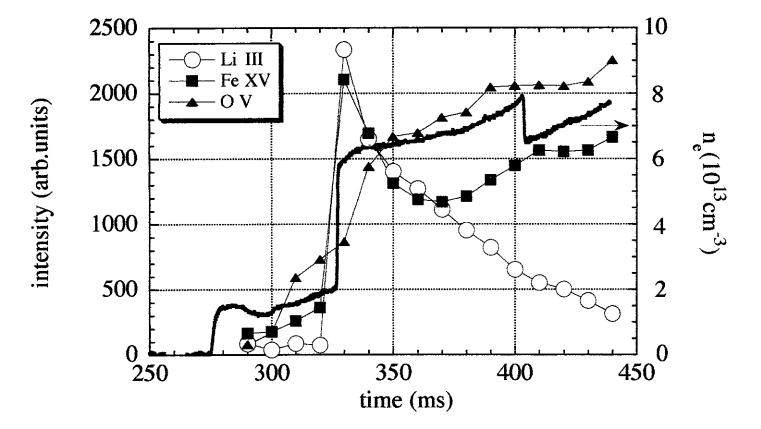


Fig.4

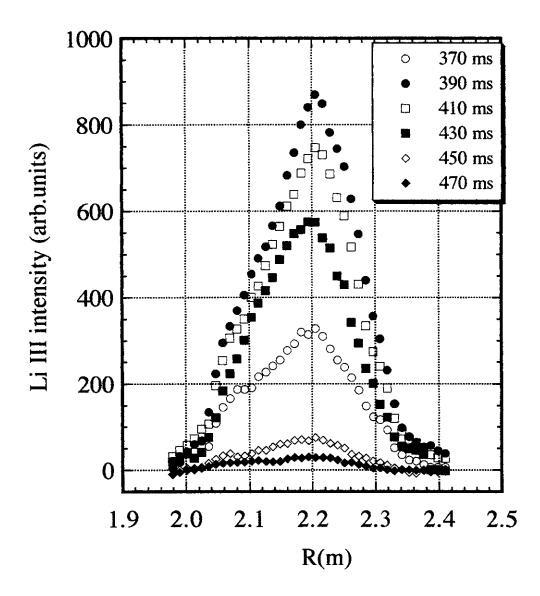


Fig.5

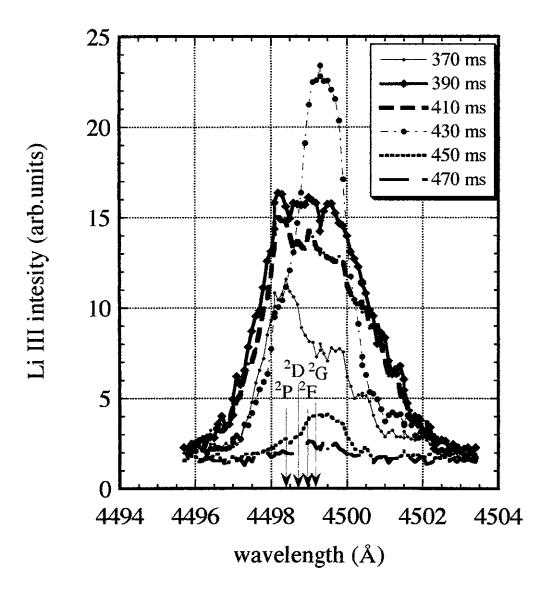


Fig.6

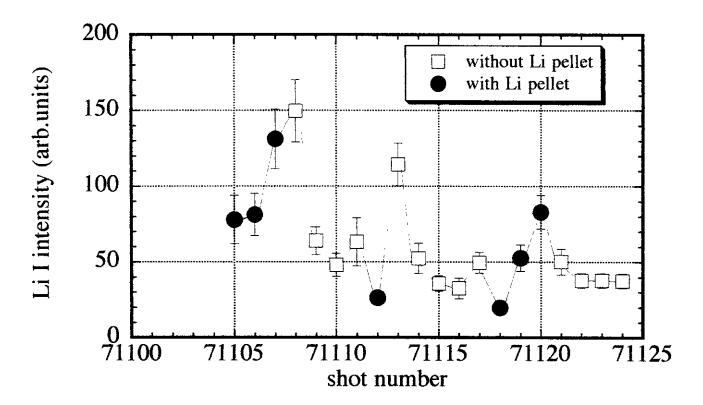


Fig.7

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