

§29. Design of Tritium Recovery System for Laser Fusion Reactor

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A tritium (T) recovery system for a Laser fusion reactor called KOYO-fast of 1GWt power is designed within the present collaboration framework between Osaka University and Kyushu University. A wetted wall of liquid $\text{Li}_{0.17}\text{Pb}_{0.83}$ eutectic alloy is proposed in order to suppress radiation damage to the first wall and to recover heat generated by neutron moderation. The LiPb eutectic alloy works not only as a protector of the first wall but also as a T-breeding blanket through the n-Li reaction. Thus LiPb works as a multi-purpose blanket fluid for T breeding, coolant and energy conversion. The outside LiPb loop is operated under the conditions of temperature and LiPb flow rate as shown in Fig. 1. Generated T is removed by a T removal system composed of a LiPb-He counter-current extraction tower. Then heat transferred to coolant is converted to electricity by a steam generator and steam turbine.

In FY 2008, experiment based on the permeation method has been performed to determine the values of permeability (P_H or P_D), solubility (K_H or K_D) and diffusivity (D_H or D_D) of H_2 or D_2 in $\text{Li}_{0.17}\text{Pb}_{0.83}$. Fig. 2 shows data of permeability of H_2 through LiPb. The pressure dependence is found to be square root of upstream pressure regardless of temperature when the downstream pressure is very low. Then, the H permeation flux, j_H , is correlated as follows:

$$j_{\text{H}_2} = \frac{1.8 \times 10^{-9}}{l} \exp\left(\frac{30.1[\text{kJ/mol}]}{R_g T}\right) \left(p_{\text{H}_2, \text{up}}^{0.5} - p_{\text{H}_2, \text{down}}^{0.5}\right) [\text{mol/m}^2\text{s}]$$

The values of solubility and diffusivity determined experimentally are presented in the papers cited below [1-3], and are correlated to the equations:

$$D_{\text{LiPb-H}} = 1.8 \times 10^{-8} \exp\left(\frac{-11.6[\text{kJ/mol}]}{R_g T}\right) [\text{m}^2/\text{s}]$$

$$K_{\text{LiPb-H}} = 2.1 \times 10^{-6} \exp\left(\frac{-18.7[\text{kJ/mol}]}{R_g T}\right) [1/\text{Pa}^{0.5}]$$

The H-D isotopic difference determined is $K_D=1.4K_H$ for solubility, $D_D=D_H$ for diffusivity and $P_D=1.4P_H$ for permeability. Based on the solubility data, the conditions necessary for the T recovery system are determined as the outlet T concentration of 5×10^{-5} ppm and the equilibrium T pressure of 2.6×10^{-7} Pa.

The T removal system is composed of an extraction tower packed with steel Rashig Ring of 1inch, and LiPb and He flow in a counter-current way. The tower height and the tower diameter are 7m and 4m. The flow rates of He and LiPb in the tower are $47 \text{ m}^3/\text{s}$ and $3.5 \text{ m}^3/\text{s}$. These conditions can achieve 10^5 of inlet/outlet T concentration ratio in LiPb.

Papers published in FY2008

(1) Y. Maeda, S. Fukada, Y. Edao, "Solubility, diffusivity and isotopic exchange rate of hydrogen isotopes in Li-Pb", Fusion Science and Technology, 54 (2008) 131-134.

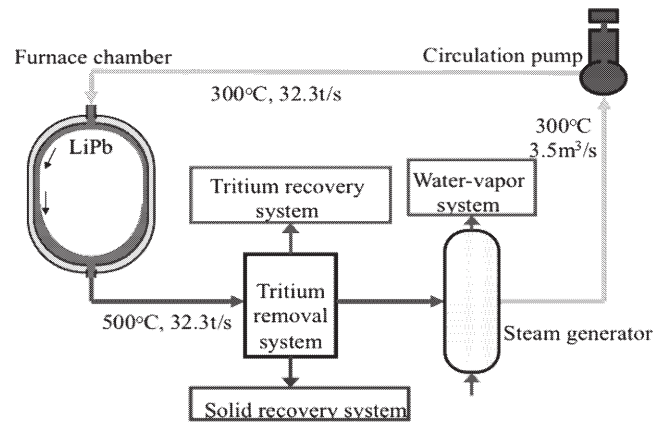


Fig. 1 LiPb circulation system for KOYO fast

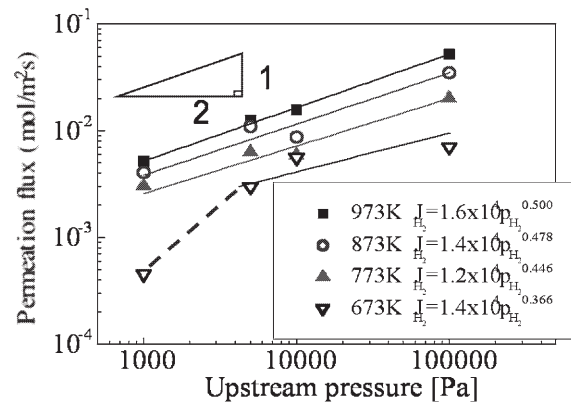


Fig. 2 Pressure dependence of hydrogen permeability through $\text{Li}_{17}\text{Pb}_{83}$

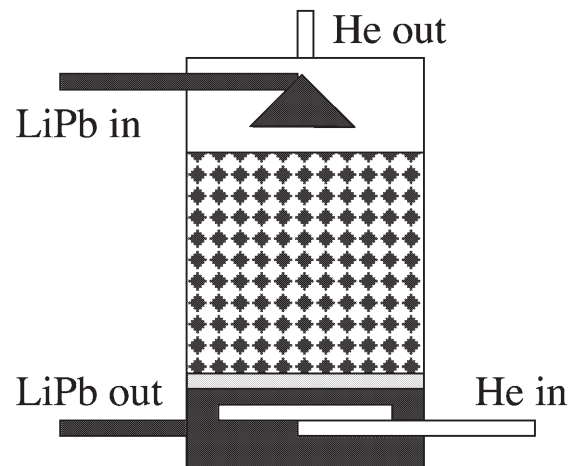


Fig. 3 LiPb-He counter-current extraction tower

(2) S. Fukada, Y. Edao, Y. Maeda, T. Norimatsu, "Tritium recovery system for Li-Pb of Inertial Fusion Reactor", Fusion Engineering and Design, 83 (2008) 747-751.

(3) S. Fukada, Y. Edao, S. Yamaguchi, "Tritium recovery from Li-Pb eutectic alloy blanket", Proc. of 2nd Japan-China workshop on blanket and tritium technology, May 9-10 (2008) 117-120.