

## §1. Increase of MgB<sub>2</sub> Phase Volume Fraction by the Reduce Particle Sized Boron Powder

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### 1. Introduction and Motivation

MgB<sub>2</sub> has higher critical transition temperature ( $T_c$ ) of 39 K compared with the other metallic superconducting compounds. Furthermore, the features of MgB<sub>2</sub> compound are not only higher  $T_c$  but also binary chemical composition, lower specific gravity and lower cost. In addition, it is well known that MgB<sub>2</sub> compound is one of the low activation superconducting materials. The induced radio-activity of MgB<sub>2</sub> compound remarkably lower than that of Nb-based superconducting materials, and the half-life period of MgB<sub>2</sub> compound was estimated within 1 year. MgB<sub>2</sub> will be one of the attractive superconducting materials to use under the higher neutron irradiation and flux environment applications such as future fusion reactor, and then it will be alternative material of Nb-Ti wire in these field applications. However, critical current density ( $J_c$ ) of MgB<sub>2</sub> wire was remarkably lower than Nb-Ti wire, and  $J_c$  enhancement is required to apply for the fusion application. In previous studies, the secondary (impurity and non-reactive) phase and voids were observed in MgB<sub>2</sub> matrix after the heat treatment, and then these are the lowering factors of  $J_c$  property. In order to improve  $J_c$  property by microstructure control of MgB<sub>2</sub> matrix, the reduce particle sized metal boron powder was prepared as the raw boron material on the in-situ MgB<sub>2</sub> wire.

### 2. Sample and reduce particle sized boron powder preparations

Precursor mixture powders were made by metal Mg powder (99.9%), Mg<sub>2</sub>Cu compound and conventional natural boron powder. The Cu additional composition of precursor powder was fixed to the optimum 3 at%Cu. The reduce particle sized boron powder was made using the high-speed vibrated milling. Boron powder and tungsten carbide (WC) ball were set into the WC pot. When boron powder was set into the WC pot, liquid nitrogen was also used in order to drive off the air gas. The milling time was 1, 3 and 5 minutes. The particle size distributions of some reduce particle sized boron powder are shown in fig.1. The particle size distributions of the boron powder after various vibrated milling time are shown in Fig. 1. The particle size distribution of conventional boron powder had two main peaks around 1.5 and 0.5  $\mu\text{m}$ , and then average particle size was obtained to 1.07  $\mu\text{m}$ . In the case of the atomized boron powder using high-speed vibrated milling during 1 minute, the particle size distribution was also shown two main peaks around 0.5 and 0.2  $\mu\text{m}$ , and then average particle size was obtained to 0.22  $\mu\text{m}$ , which value was corresponded to about 1/5 compared with conventional boron powder. These mentioned that the particle size of boron powder could be

reduced by the high-speed vibrated milling. According to the XRD analysis of all atomized boron powders, the diffraction peaks of B<sub>2</sub>O<sub>3</sub> phase were not observed at all. The liquid N<sub>2</sub> into WC pot was effective to restrain the boron oxidation.

### 3. Increase of MgB<sub>2</sub> phase volume fraction by reduce particle sized boron powder

Typical magnetization curves of MgB<sub>2</sub> wires using reduce particle sized boron powder as a function of the vibrated milling time is shown in Fig. 2. The magnetic moment was normalized by the total volume of wire filaments.  $T_c$  values of MgB<sub>2</sub> wire using reduce particle sized boron powder were almost obtained to 36.5 K and they were same to the MgB<sub>2</sub> wire using conventional boron powder. On the other hands, magnetization moment which was normalized by the total core volume was increased by the reduce particle sized boron powder. This suggested that the volume fraction of MgB<sub>2</sub> phase was increased and reduce particle sized boron promoted to form MgB<sub>2</sub> phase.

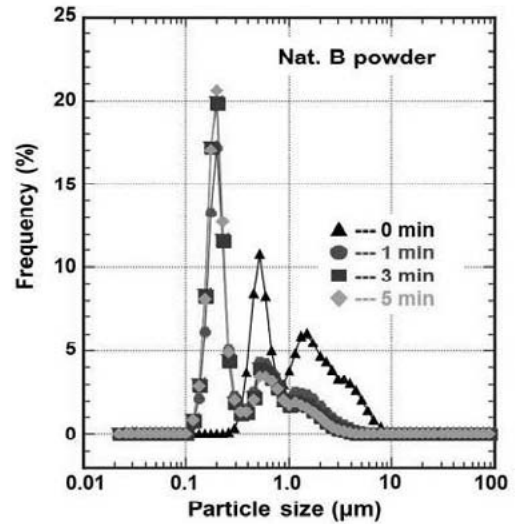


Fig.1 Typical particle size distributions of the boron powder after various vibrated milling time

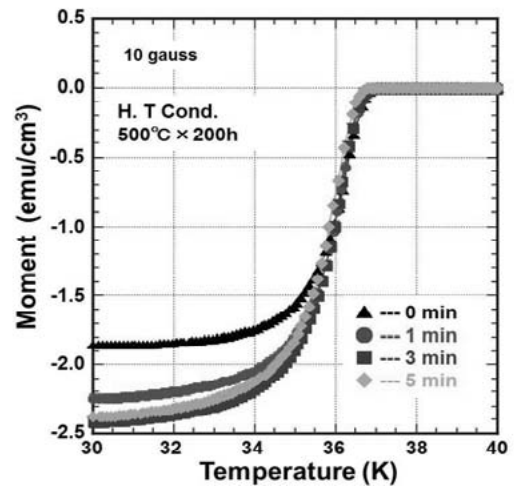


Fig.2 Typical magnetization curves of MgB<sub>2</sub> wires using reduce particle sized boron powder as a function of the vibrated milling time